

Atomic Basics Structure Answers

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Basic atomic structure worksheet h and the 1. 2. 4. 5. 6. 8. 9. the 3 particles of the atom are: protons (p) (o their respective charges are: the number of protons in one atom of an element determines the atom's number of electrons determines the of the element in one atom of an element. it also in a neutral atom of that element. the atomic number

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Answer: an atom contains subatomic particles called protons, neutrons, and electrons. explanation: the basic structure of an atom includes a tiny, relatively massive nucleus, containing at least one proton and usually one or more neutrons.

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Basic atomic structure worksheet answers 1 a) protons b) neutrons c) electrons a) positive b) neutral c) negative 2 atomic number (or identity); charge 3 protons; electrons (in a neutral charge atom only!); same 4 average atomic weight; mass 5 mass number; nucleus 6 neutrons; protons (or atomic number); mass number 7 lithium = li = 3 bromine = br = 35 iron = fe = 26 copper = cu = 29 oxygen = o = 8 mercury = hg = 80 krypton = kr = 36 helium = he = 2 8 (a neutral atom by definition has no charge!

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